Highly Concise and Stereoselective Total Synthesis of (5R,7S)-Kurzilactone

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A highly concise and stereoselective total synthesis of (5R,7S)-kurzilactone (1) was performed by a convergent approach by means of a *Jacobsen's* hydrolytic kinetic resolution, a Hor*ner–Wadsworth–Emmons* reaction for the construction of the α , β -unsaturated δ -lactone ring system, and a highly diastereoselective Mukaiyama aldol reaction for the introduction of the formal anti-1,3-diol unit (Schemes 2 and 3).

Introduction. – Natural products possessing 6-substituted α , β -unsaturated- δ lactone moieties have attracted the attention of synthetic organic chemists due to their antitumor properties [1]. In addition, they inhibit HIV proteases [2], induce apopotosis [3] [4], and were shown to be antileukemic [5], along with having many other relevant pharmacological properties [6]. (5R,7S)-Kurzilactone¹) (1), an α , β unsaturated δ -lactone, was isolated from the leaves of *Cryptocarya kurzii*, a plant that is indigenous to Malaysia. Kurzilactone showed a marked cytotoxicity against the KB human-carcinoma cell line $(IC_{50} = 1 \text{ µg/ml})$ [7]. Initially, the stereogenic centers bearing the OH groups in the side chain and the O-atom of the δ -lacton were assigned a syn-relationship through NMR experiment but a corrected *anti*-relationship with $(5R,7S)$ configuration of the C(5) and C(7) stereogenic centers were later assigned on the basis of a total synthesis [8]. The structural uniqueness of kurzilactone, coupled with its interesting bioactivity and our interest to verify the selectivity of the Mukaiyama aldol reaction in case of a δ -lactone-substituted aldehyde instead of an open-chain aldehyde as reported by our group, prompted us to revisit its total synthesis. The retrosynthetic analysis of 1 is described in *Scheme 1*.

Results and Discussion. – Our initial approach to aldehyde 8 employed the oxirane **2** [9] (*Scheme 2*), prepared by *Jacobsen'*s hydrolytic kinetic resolution (HKR) [10] of the racemate (*Scheme 2*) in the presence of $\left[\text{Co}^{\text{III}}\{(R,R)-\text{Salen}\}(OAc)\right]$ catalyst $('salen' = N, N'-bis[3,5-di(tert-buty]) salicylidene] cyclohexane-1,2-diaminato = [{2,2'-}$ $[$ (cyclohexane-1,2-diylbis(nitrilomethylidyne)]bis[4,6-di(tert-butyl)phenylato]}(2-). The 4-methoxybenzyl(PMB)-protected $(2R)$ -oxirane 2 was obtained as a single enantionmer ($\left[\alpha\right]_D^{25} = +10.6$ (c = 1.2, CHCl₃), ($\left[11\right]$: $\left[\alpha\right]_D^{25} = -13.1$ (c = 1.2, CHCl₃) for the (2S)-enantiomer)), which was easily separated from the more polar diol. Oxirane 2 was treated with vinylmagnesium bromide in the presence of CuI to afford

¹) Trivial atom numbering; for the systematic name, see *Exper. Part.*

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Scheme 1. Retrosynthetic Analysis of (5R,7S)-Kurzilactone (1)

the homoallyl alcohol 3 in 85% yield. The spectral and analytical data of 3 (ee 94% by $HPLC$) were in good agreement with the literature values [12]. The $C(3)$ stereogenic center was further confirmed by a modified *Mosher'*s method [13]. $(3S)$ -Alcohol **3** was esterified with 2-(diethoxyphosphinyl)acetic acid 4 [14] in the presence of dicyclohexylcarbodiimide (DCC) and a catalytic amount of N,N-dimethylpyridin-4-amine (DMAP) to afford the (S) -ester 5 in 82% yield. Removal of the PMB group with $2,3$ -dichloro-5,6-dicyano-1,4-benzoquinone $(=4,5$ -dichloro-3,6-dioxocyclohexa-1,4-diene-1,2-dicarbonitrile; DDQ) [15] in CH_2Cl_2 yielded hydroxy derivative 6 in 90% yield. Oxidation of 6 with 2-iodoxybenzoic acid (IBX) [16] followed by a *Horner–Wadsworth–Emmons* reaction provided $\alpha,\!\beta$ -unsaturated δ -lactone **7** in 68% yield over two steps. The one-step conversion of the terminal olefin moiety of 7 to give aldehyde 8 was achieved by a modified dihydroxylation followed by oxidative cleavage of the diol in 91% yield [17].

Scheme 2. Synthesis of Aldehyde 8

a) Vinylmagnesium bromide, CuI THF, 0° to r.t., 2 h; 85%. b) (EtO)₂P(O)CH₂COOH (4), DCC, DMAP, CH₂Cl₂, r.t., 12 h; 62%. c) DDQ, CH₂Cl₂, H₂O, 0°, 2 h; 90%. d) 1. IBX, DMSO, THF, 0° to r.t., 6 h; 2. NaH, THF, 0° , 2 h; 68% over two steps. e) OsO₄, 2,6-lutidine (=2,6-dimethylpyridine), NaIO₄, dioxane, H₂O, r.t., 4 h; 91%.

Having aldehyde 8, the stage was set to carry out the crucial Mukaiyama aldol reaction to verify the selectivity. The $BF_3 \cdot Et_2O$ -mediated *Mukaiyama* aldol reaction of aldehyde 8 with trimethylsilyl enol ether 9 derived from $(3E)$ -4-phenylbut-3-en-2-one afforded the aldol adduct $(5R,7S)$ -kurzilactone (1) as the only product in 95% yield (Scheme 3). It is noteworthy to mention here that in case of the α , β -unsaturated- δ lactone-substituted aldehyde 8, the cyclic transition state favored the attack from the opposite face of the lactone center leading to exclusive (\geq 99%) formation of (5R,7S)kurzilactone (1) . The spectral and analytical data of the synthetic $(5R,7S)$ -kurzilactone (1) were in good agreement with the reported value [8].

Scheme 3. Synthesis of (5R,7S)-Kurzilactone (1)

a) Et₃N, CF₃SO₃SiMe₃, CH₂Cl₂, -10° , 30 min. b) **8**, BF₃ · Et₂O, CH₂Cl₂, -78° , 2 h; 81% over two steps.

Conclusions. – We achieved the total synthesis of $(5R,7S)$ -kurzilactone (1) in five longest linear steps starting from a known oxirane by means of a *Horner*-*Wadsworth–Emmons* reaction for the construction of the α , β -unsaturated δ -lactone ring system and a highly diastereoselective Mukaiyama aldol reaction for the introduction of the formal *anti*-1,3-diol unit. Following the same protocol, $(5S,7R)$ kurzilactone (ent-1) could also be synthesized.

Experimental Part

General. Air- and/or moisture-sensitive reactions were carried out in anh. solvents under Ar in an oven- or flame-dried glassware. All anh. solvents were distilled prior to use: THF, benzene, toluene and Et₂O from Na and benzophenone; CH₂Cl₂, quinoline, and Et₃N from CaH₂; MeOH from Mg cake. Commercial reagents were used without purification. Column chromatography (CC) : silica gel $(SiO₂)$, 60–120 mesh). Specific optical rotations: $\lbrack \alpha \rbrack_{D}$ in 10^{-1} deg cm² g⁻¹. IR Spectra: in CHCl₃ or neat (as mentioned); $\tilde{\nu}$ in cm⁻¹. ¹H- and ¹³C-NMR Spectra: at 300 (¹H) and 75 MHz (¹³C); in CDCl₃; δ in ppm rel. to Me4Si as internal standard, J in Hz.

 $(3S)-1-(4-Methoxyphenyl/methoxyJhex-5-en-3-ol (3).$ To a stirred soln. of 2 (5.0 g, 24.22 mmol) and CuI (0.45 g, 2.40 mmol) in THF (30 ml) was added CH₂=CHMgBr (48 ml, 48.02 mmol) at -10° , and stirred at -10° for 1 h. After completion of the reaction (TLC monitoring), it was quenched with sat. NH₄Cl soln. (20 ml). The mixture was extracted with AcOEt (2×60 ml), the combined org. layer washed with H₂O (100 ml) and brine (100 ml), dried (Na₂SO₄), and concentrated, and the residue purified by CC (SiO_2) : **3** (4.8 g, 85%). Light yellow liquid. R_f (AcOEt/hexane 3:7) 0.4. $[\alpha]_D^{25} = -6.2$ (c = 1.1, CHCl₃) $([12]$: $[\alpha]_D^{20} = -5.8$ (c = 1.37, CHCl₃)). IR (KBr): 2923, 2855, 2380, 1724, 1611, 1512, 1247, 1096, 1034, 822, 560. ¹H-NMR: 7.25 (d, J = 8.4, 2 H); 6.87 (d, J = 8.4, 2 H); 5.80 – 5.72 (m, 1 H); 5.14 – 5.06 (m, 2 H); 4.45 $(s, 2 H)$; 3.89 – 3.82 $(m, 1 H)$; 3.80 $(s, 3 H)$; 3.73 – 3.57 $(m, 2 H)$; 2.27 – 2.20 $(m, 2 H)$; 1.82 – 1.70 $(m, 2 H)$; 13C-NMR: 159.1; 134.8; 129.9; 129.2; 117.4; 113.7; 72.8; 70.3; 68.5; 55.2; 41.8; 35.7. ESI-HR-MS: 259.1305 $([M+Na]^+, C_{14}H_{20}NaO_3^+;$ calc. 259.1306).

(1S)-1-{2-[(4-Methoxyphenyl)methoxy]ethyl}but-3-en-1-yl 2-(Diethoxyphosphinyl)acetate (5). To a stirred soln. of 2-(diethoxyphosphinyl)acetic acid (4; 4.5 g, 19.07 mmol) and DCC (7.9 g, 38.14 mmol) in CH_2Cl_2 (50 ml), was added DMAP (1.1 g, 9.05 mmol) at r.t. The mixture was stirred for 15 min before a soln. of 3 (7.4 g, 38 mmol) in CH₂Cl₂ (30 ml) was added at r.t. The mixture was stirred at r.t. for 12 h. After completion of the reaction (TLC monitoring), the mixture was quenched with $H₂O$ (30 ml). The

aq. layer was extracted with CH₂Cl₂ (2×75 ml), the combined org. phase dried (Na₂SO₄), and concentrated, and the residue purified by CC (SiO₂): **5** (4.5 g, 62%). Colorless liquid. R_f (AcOEt/hexane 1:1) 0.3. $\lbrack a \rbrack_5^2 = +20.4$ (c = 1.5, CHCl₃). IR (KBr): 3328, 2926, 2853, 1666, 1734, 1625, 1250, 1029, 966, 610. ¹H-NMR: 7.24 (d, J = 8.4, 2 H); 6.86 (d, J = 8.4, 2 H); 5.73 – 5.69 (m, 1 H); 5.17 – 5.04 (m, 3 H); 4.39 $(s, 2 H)$; 4.20 – 4.08 (m, 4 H); 3.82 – 3.76 (s, 3 H); 3.57 – 3.43 (m, 2 H); 2.92 (d, J = 21.5, 2 H); 2.42 – 2.27 $(m, 2 H)$; 1.91 – 1.80 $(m, 2 H)$; 1.34 $(dd, J=6.97, 0.37, 6 H)$. ¹³C-NMR: 164.4; 158.3; 132.4; 129.6; 128.5; 117.3; 112.9; 71.8; 71.4; 65.2; 61.7; 54.4; 37.8; 34.4; 33.2; 15.6. ESI-HR-MS: 437.1705 ($[M + Na]$ ⁺, $C_{20}H_{32}NaO_7P^+$; calc. 437.1726).

(1S)-1-(2-Hydroxyethyl)but-3-en-1-yl 2-(Diethoxyphosphinyl)acetate (6). To a stirred soln. of 5 $(4.5 g, 10.81 mmol)$ in CH₂Cl₂/H₂O 9:1 (50 ml) was added DDO (2.9 g, 13.12 mmol) at 0°, and the mixture was stirred for 2 h. After completion (TLC monitoring), the reaction was quenched with sat. aq. NaHCO₃ soln. (25 ml). The mixture was filtered through *Celite*, the aq. layer extracted with CH₂Cl₂ (2 \times 40 ml), the combined org. phase washed with H₂O (50 ml) and brine (50 ml), dried (Na₂SO₄), and concentrated, and the residue purified by CC (SiO₂): 6 (2.87 g, 90%). Colorless liquid. R_f (AcOEt/hexane $4:1)$ 0.2. $\lbrack a \rbrack_2^2 = +10.6$ $(c = 1.7, CHCl_3)$. IR (KBr): 2922, 2854, 2381, 1733, 1458, 1271, 1030, 970, 604, 494. 1 H-NMR: 5.79 – 5.72 (m, 1 H); 5.16 – 5.04 (m, 3 H); 4.23 – 4.10 (m, 4 H); 3.72 – 3.57 (m, 2 H); 3.02 – 2.80 $(m, 2 H); 2.44 - 2.32 (m, 2 H); 1.89 - 1.80 (m, 1 H); 1.75 - 1.66 (m, 1 H); 1.40 - 1.33 (m, 6 H).$ ¹³C-NMR: 165.9; 133.0; 118.0; 72.7; 63.0; 58.2; 38.7; 36.2; 33.5; 16.2. ESI-HR-MS: 295.1310 ($[M + H]^+, C_{12}H_{24}O_6P^+;$ calc. 295.1305).

(6S)-5,6-Dihydro-6-(prop-2-en-1-yl)-2H-pyran-2-one (7). To a stirred soln. of iodoxybenzoic acid (3.7 g, 13.22 mmol) in dry DMSO (10 ml), was added a soln. of 6 (2.6 g, 8.84 mmol) in anh. THF (35 ml) at r.t. and stirred for 6 h. After completion of the reaction (TLC monitoring), the mixture was quenched with H₂O (20 ml) and filtered through *Celite*. The filtrate was extracted with AcOEt (3×50 ml). The combined org. phase was washed with brine (75 ml), dried ($Na₂SO₄$), and concentrated, and the resulting crude aldehyde (2.75 g) was used as such without further purification for the next step. To a suspension of NaH (0.32 g, 13.62 mmol) in dry THF (60 ml), the crude aldehyde (2.75 g, 8.06 mmol) in THF (200 ml) was added at 0° , and the mixture was stirred for 2 h. After completion of the reaction (TLC monitoring), it was quenched with H₂O (30 ml). THF was removed, the aq. layer extracted with AcOEt (3×50 ml), the combined org. phase dried (Na_2SO_4) and concentrated, and the residue purified by CC (SiO₂): 7 $(0.64 \text{ g}, 68\% \text{ over two steps})$. Colorless liquid. R_f (AcOEt/hexane 3:7) 0.7. $[a]_D^{25} = +84$ (c = 2.1, CHCl₃). IR (KBr): 2923, 2855, 1726, 1459, 1280, 1071, 526. ¹H-NMR: 6.87 – 6.79 (m, 1 H); 6.00 (dt, J = 9.8, 1.5, 1 H); 5.88 – 5.79 (m, 1 H); 5.20 – 5.11 (m, 2 H); 4.49 – 4.42 (m, 1 H); 2.62 – 2.40 (m, 2 H); 2.37 – 2.31 (m, 2 H). ¹³C-NMR: 164.2; 144.9; 132.2; 121.2; 118.8; 77.0; 39.0; 28.6. ESI-MS: 138 (M⁺).

 $(2R)$ -3,6-Dihydro-6-oxo-2H-pyran-2-acetaldehyde (8). To a stirred soln. of 7 (0.2 g, 1.42 mmol) in dioxane/H₂O 3:1 (8 ml), 2,6-lutidine (0.31 g, 2.89 mmol), OsO₄ (0.007 g, 0.028 mmol), and NaIO₄ (1.23 g, 5.79 mmol) were added. The mixture was stirred at r.t. for 4 h (TLC monitoring). The mixture was quenched with H₂O (10 ml) and extracted with CH₂Cl₂ (3 \times 30 ml). The combined org. phase was washed with brine (50 ml), dried (Na₂SO₄), and concentrated, and the resulting crude 8 (208 mg, 91%) was used as such without further purification in the next step.

 $\frac{f}{IE}$ -3- $\frac{f}{T}$ (Trimethylsilyl)oxy]buta-1,3-dien-1-yl}benzene (9). To a stirred soln. of (3E)-4-phenylbut-3-en-2-one $(0.5 \text{ g}, 3.44 \text{ mmol})$ and $Et_3N (0.47 \text{ ml}, 5.17 \text{ mmol})$ in freshly prepared anh. $CH_2Cl_2 (10 \text{ ml})$ at -10° was added dropwise CF₃SO₃SiMe₃ (0.8 ml, 4.10 mmol). The mixture was stirred at -10° for 30 min and quenched with sat. aq. NaHCO₃ soln. The aq. layer was extracted with CH₂Cl₂ (2×25 ml), the combined org. phase dried (Na₂SO₄) and concentrated at r.t., and the resulting crude 9 (0.46 g) was used as such for the next step without purification.

 $(6R)$ -5,6-Dihydro-6- $[(2S,5E)$ -2-hydroxy-4-oxo-6-phenylhex-5-en-1-yl]-2H-pyran-2-one (1). Aldehyde 8 (0.208 g, 1.4 mmol) and silyl enol ether 9 (0.46 g, 2.12 mmol) were taken up in freshly prepared anh. CH₂Cl₂ (20 ml), and the mixture was cooled to -78° . BF₃ · Et₂O (0.30 ml, 2.12 mmol) in freshly prepared anh. CH₂Cl₂ (5 ml) was added slowly. The mixture was stirred at -78° for additional 2 h and quenched with sat. aq. NaHCO₃ soln. (10 ml). The the aq. layer was extracted with CH₂Cl₂ (2 \times 25 ml), the combined org. phase washed with H₂O (40 ml) and brine (40 ml), dried (Na₂SO₄), and concentrated, and the residue purified by CC (SiO₂): **1** (382 mg, 81% over two steps). White solid. R_f (AcOEt/hexane 1:3) 0.3. M.p. 73–74°. $\lbrack \alpha \rbrack_{D}^{25} = +82.4 \quad (c = 1.0, \text{ CHCl}_3) \quad (\lbrack 8 \rbrack; \lbrack \alpha \rbrack_{D}^{20} = +84 \quad (c = 0.231, \text{ CHCl}_3)$. IR (KBr):

3453, 3028, 2924, 2854, 1712, 1604, 1384, 1252, 1049, 781, 508. ¹H-NMR: 7.58 (d, J = 16.0, 1 H); 7.56 – 7.52 $(m, 2H)$; 7.43– 7.36 $(m, 3H)$; 6.94–6.91 $(m, 1H)$; 6.77–6.68 $(d, J=16.0, 1H)$; 6.02 $(dd, J=9.8, 2.0$, 1 H); 4.85 – 4.75 (m, 1 H); 4.54 – 4.46 (m, 1 H); 2.93 (dd, J = 17.3, 3.0, 1 H); 2.81 (dd, J = 17.3, 9.0, 1 H); 2.44 – 2.30 (m, 2 H); 1.93 – 1.78 (m, 2 H). 13C-NMR: 200.3; 164.2; 145.2; 143.8; 134.0; 130.8; 128.9; 128.4; 125.9; 121.3; 74.8; 63.9; 46.8; 41.6; 29.9. ESI-HR-MS: 309.1102 $([M + Na]^+, C_{17}H_{18}NaO_4^+;$ calc. 309.1109).

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